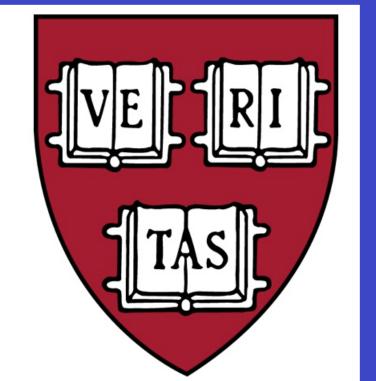


Influence of Fires on Air Quality During the SEAC⁴RS 2013 Campaign

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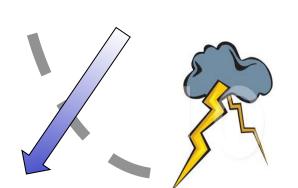


1. Motivation

Ground-level O₃ pollution is a serious public health and environmental concern in the U.S. I am interested in improving understanding of background O₃, a critical issue for setting the National Ambient Air Quality Standards (NAAQS). EPA is considering tightening the O₃ NAAQS, but it is unclear to what extent uncontrollable background O₃ would hinder achievement of a lower NAAQS (Macdonald-Buller et al, 2011). Addressing this issue is of critical importance for policy and presents a major scientific challenge to coupling atmospheric chemistry on global and regional scales.

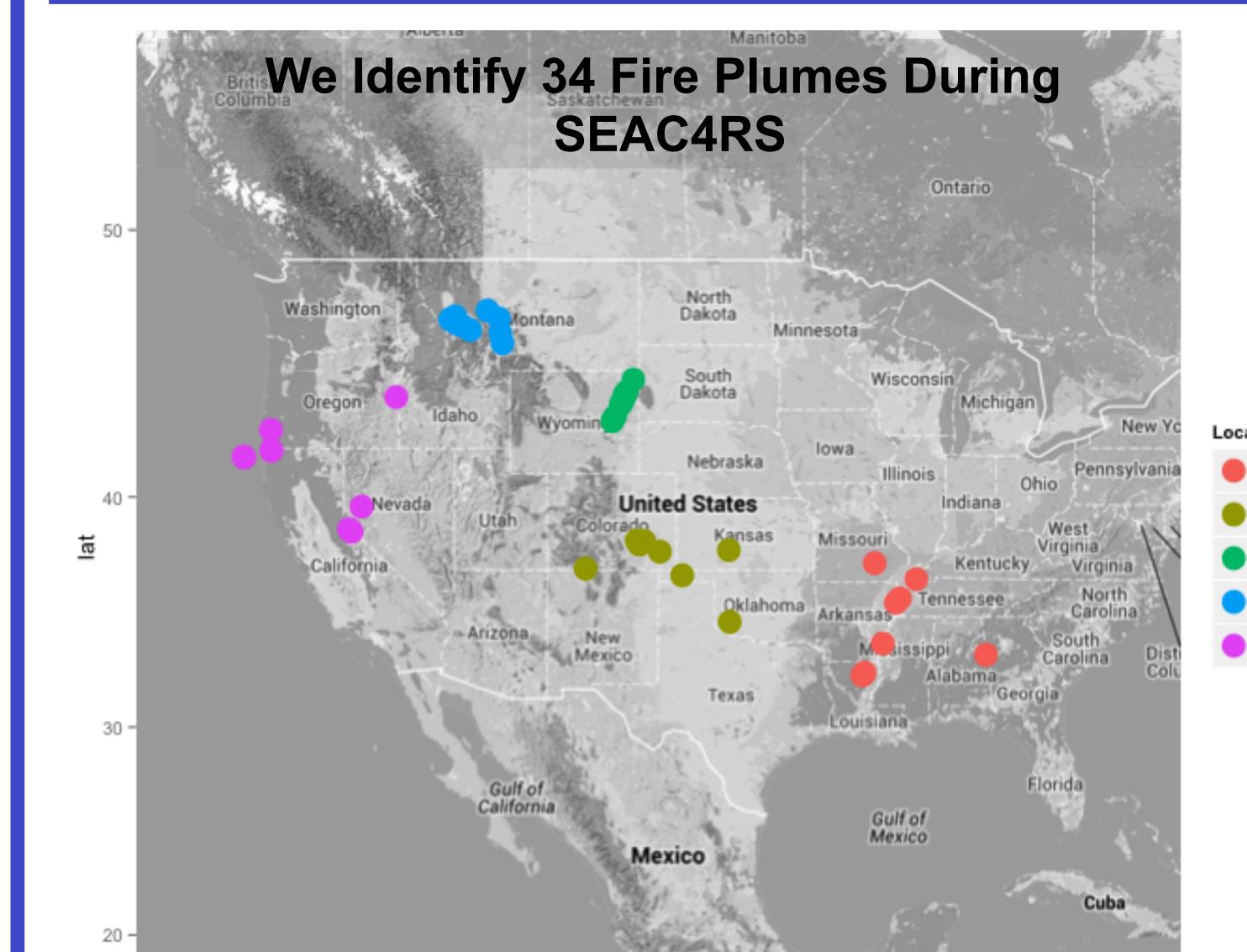
Background O₃ is not directly measurable and thus must be calculated from a global chemical transport models (CTM). Current CTMs have difficulty capturing observed high background ozone events (over 70ppb) due a variety of limitations including modeling of wildfires¹. CTMs also show large differences in their representation of the impact of wildfires on the ozone background (Fiore et al, 2009) which is very problematic for air quality policy.







2. Fires Sampled During SEAC⁴RS



Plumes with high $\Delta O_3/\Delta CO$ have enhanced PAN. PAN may be an important NOx reservoir for downwind O3 production as well. Thus successful modeling of fire impacts on background ozone requires successful modeling of PAN.

Mean $\Delta O_3/\Delta CO$ – ppbv/ppbv

	SE	Central	Wyoming	NW	California
km	(n=5)	(n=8)	(n=5)	(n=9)	(n=7)
0-1	0.0	_	-	-	-
1-2	0.0	_	-	_	0.02
2-3	_	0.08	-	_	0.02
3-4	0.04	0.09	-0.02	-0.02	0.02
4-5	0.15	0.07	0.13	0.01	0.07
5-6	_	0.05	-0.01	_	_

Mean ΔPAN/ΔCO – pptv/ppbv

	SE	Central	Wyoming	NW	California
km	(n=5)	(n=8)	(n=5)	(n=9)	(n=7)
0-1	-0.7	_	_	_	_
1-2	-0.9	_	_	_	2.6
2-3	_	7.8	_	_	3.9
3-4	5.5	9.1	3.6	4.6	4.1
4-5	6.8	5.0	9.1	4.4	4.5
5-6	—	5.6	6.3		_

3. Geos-Chem NOx-O3-VOC Simulation

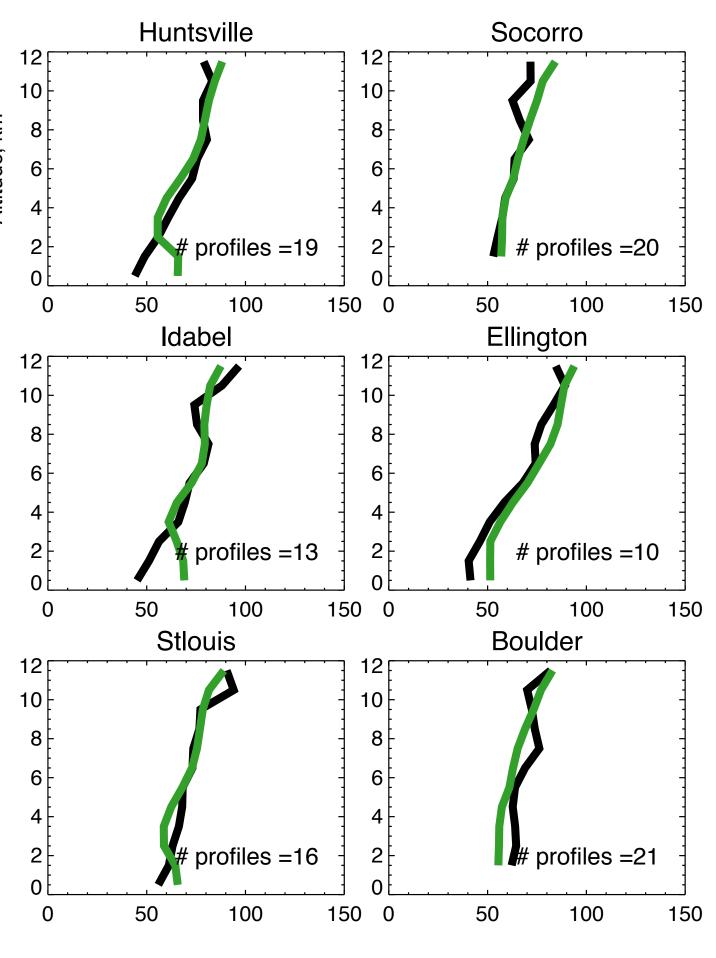
- Resolution = $0.25^{\circ} \times 0.3125^{\circ}$ and $4^{\circ} \times 5^{\circ}$
- Meteorology = GMAO GEOS-FP
- Emissions = NEI2008, FINN, MEGAN2.1
- Soil NOx = Hudman et al, 2012
- Isoprene chemistry = Mao et al, 2013 Fire Injection
- Emit 40% of NOx as PAN and 20% as HNO3 (Alvarado et al, 2010).
- Inject 15% into 3-5km in the West to match SEAC⁴RS observations.
- For the rest of the northern midlatitudes, inject 15% into the first few km of the free troposphere.



MVK+MACR [ppb]

Pl. A. Wisthaler

Geos-Chem O₃ compares well with **SEACIONS** in the free troposphere.

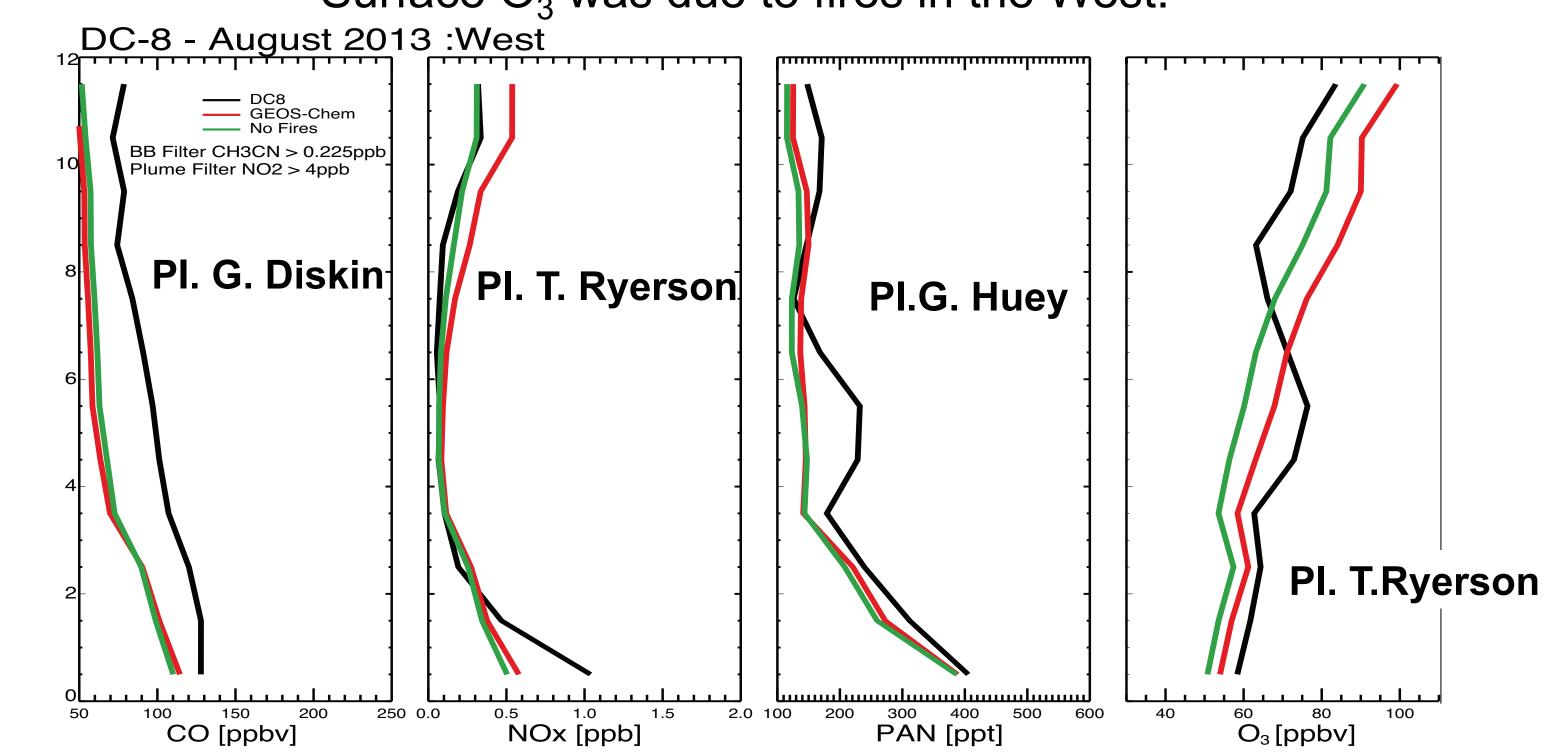


PI: Anne Thompson

With thanks to Bob Yokelson

4. Modeled Impact of Fires on Surface O₃

Geos-Chem Predicts that During the SEAC4RS Time Period, ~4ppb of Surface O₃ was due to fires in the West.



Geos-Chem overall underestimate surface ozone in the West which may be due to inadequate VOC and NOx emissions from either fires or anthropogenic sources.

Peroxyacetyl Radical Generation:

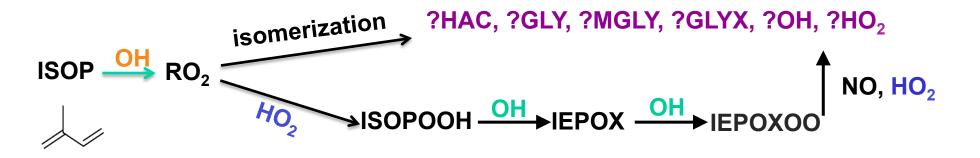
5. Future Work

Free Tropospheric O₃ in the West:

Improve modeling of near-source VOC chemistry from fires, and resultant ozone and PAN formation, in collaboration with Matt Alvarado using the ASP model.

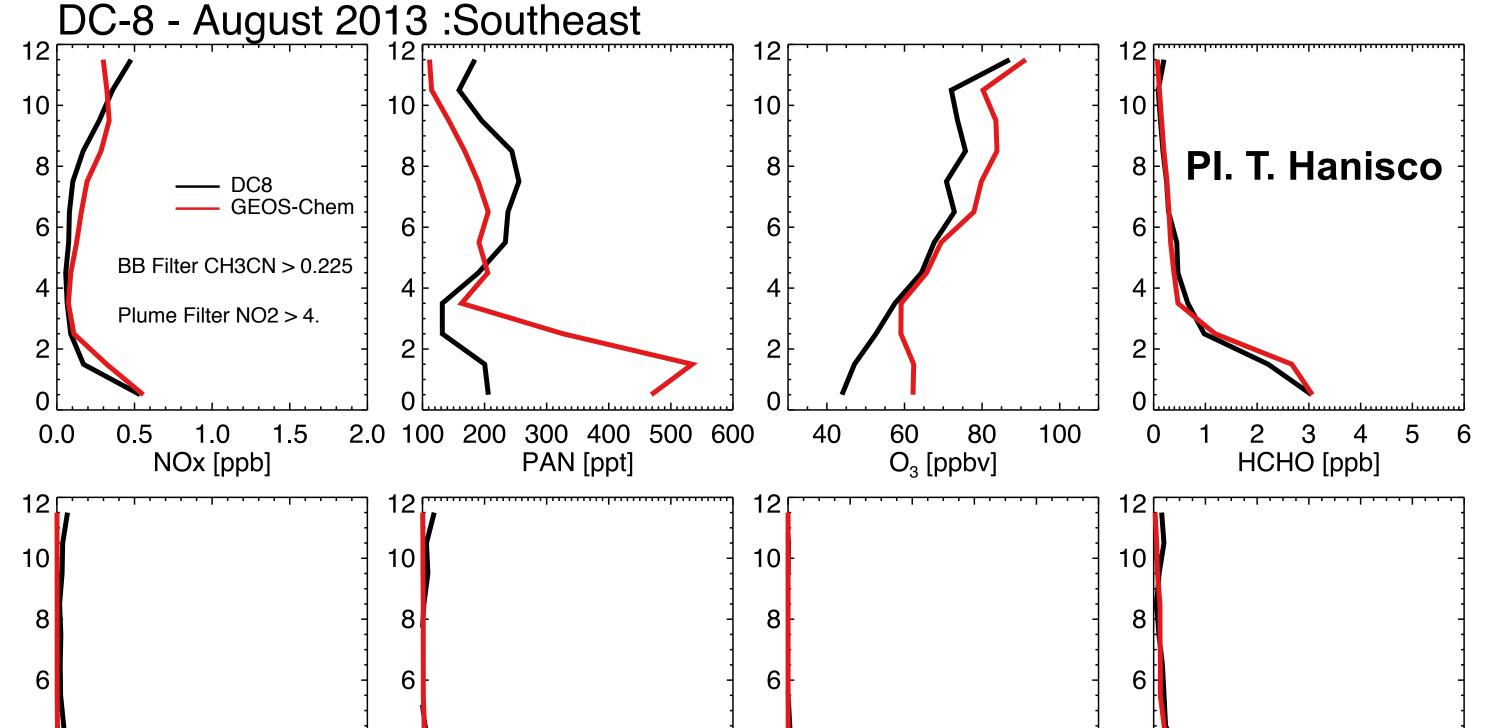
Southeast U.S. Chemistry:

- Proper treatment of new isoprene oxidation products:
 - Wet and dry deposition
 - Aerosol uptake
 - Transport
- Incorporate the most recent studies on epoxides and isomerization to test uncertainties in these mechanisms.



6. Conclusions O₃ chemistry in the southeast U.S. is sensitive to the

Fate of Isoprene Hydroperoxy Radical



ISOPOOH [pptv]

PI: P.Wennberg

100 200 300 400 500

IEPOX [pptv]

4. Modeled Southeast U.S. Chemistry

200 400 600 800 1000 0.0 0.2 0.4 0.6 0.8 1.0 HAC [ppbv]

Southeast (RIO2) - Southeast 0% 2% 0% MGLY (hv) 6% \blacksquare RIO2 + NO \blacksquare ALD2 + OH ■ RIO2 + HO2 \blacksquare MGLY + OH RIO2 Isom \blacksquare VRO2 + NO 10% 31% ■ RIO2 + RIO2 ■ VRO2 + HO2 **58%** \blacksquare RIO2 + MCO3 \blacksquare MAO3 + NO ■ RIO2 + MO2 **15%** 26% \blacksquare ISOP + O3 Other

The isomerization pathway compares well with the findings of Crounse et al, 2011. However, in the southeast, MGLY from photolysis and OH oxidation accounts for ~50% of the PA radical. MGLY comes from HAC oxidation, various parts of MVK oxidation, and directly from isoprene RO₂ oxidation. Approximately 50% of the HAC comes from isomerization as well, with 25% coming from epoxides. Therefore it is difficult to reconcile measured PAN and O₃ with the isomerization schemes currently available (Peeters et al, 2009, Crounse et al, 2011, Fuchs et al, 2013) with the epoxide chemistry adding additional uncertainty (Jacobs et al, 2013, Bates et al, 2014).

products from isomerization and epoxides. The general problem of CTMs in overestimating O₃ in the southeast and underestimating O_3 in the West persists.

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REFERENCES

Bates, K. H., et al (2014), Gas Phase Production and Loss of Isoprene Epoxydiols, Journal of Physical Chemistry A, 118(7), 1237-1246. Crounse, J. D. et al (2011), Peroxy radical isomerization in the oxidation of

isoprene, Physical chemistry chemical physics: PCCP, 13(30), 13607-13613. Fiore, A. M., et al. (2009), Multimodel estimates of intercontinental source-receptor relationships for ozone pollution, *Journal of Geophysical Research*, 114(D4). Fuchs, H., et al. (2013), Experimental evidence for efficient hydroxyl radical regeneration in isoprene oxidation, *Nature Geoscience*, 6(12), 1023-1026. McDonald-Buller, E. C., et al. (2011), Establishing policy relevant background

(PRB) ozone concentrations in the United States, Environmental science & technology, 45(22), 9484-9497. Jacobs, M.I. et al. (2013), Rate Constants and Products of the OH Reaction with

Isoprene-Derived Epoxides, Environmental science & technology, 47 12868-12876

Paulot, F., et al (2009), Unexpected epoxide formation in the gas-phase photooxidation of isoprene, Science, 325(5941), 730-733. Peeters, J., and J. F. Muller (2010), HO(x) radical regeneration in isoprene oxidation via peroxy radical isomerisations. II: experimental evidence and global

impact, Physical chemistry chemical physics: PCCP, 12(42), 14227-14235. Peeters, J., et al (2009), HOx radical regeneration in the oxidation of isoprene, Physical chemistry chemical physics: PCCP, 11(28), 5935-5939.